

ASYMMETRIC SYNTHESIS IX¹ : PREPARATION OF
CHIRAL α -SUBSTITUTED PHENETHYLAMINES

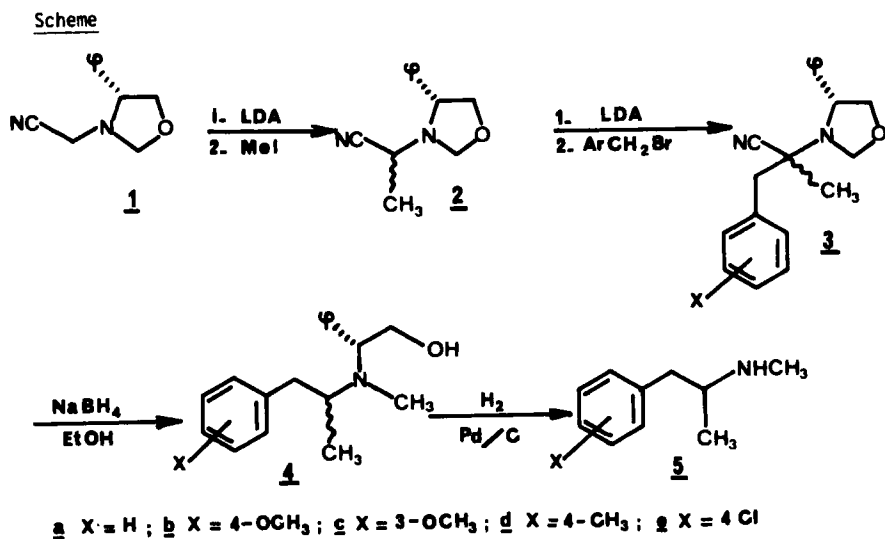
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ABSTRACT : (S)-N-methyl- α -methyl-phenethylamines 5a-d were obtained in 56-62 % e.e. from the chiral synthon (-)-N-cyanomethyl-4-phenyl-1,3-oxazolidine-1.

Despite their simple structure and important biological activities the asymmetric synthesis of amphetamines has not been the subject of many published work². The recent publication of Takahashi³ describing the synthesis of chiral 1-alkyl-2-phenylethylamines from 4-phenyl-1,3-oxazolidines prompts us to report the results of our investigations in this area.

In conjunction with our interest in preparation of optically active amines, aminoalcohols and aminoacids we have designed a new N-cyanomethyl-1,3-oxazolidine synthon 1⁴. In recent years, α -aminonitriles have been demonstrated to be useful for alkylation of the α position of the amino group leading to α -substituted amines after decyanation⁵. Our strategy for the

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Table

Diastereomeric excesses and chemical shifts of the CH₃ and N-CH₃ signals in the ¹H NMR spectra of the diastereomeric mixture of 4 (CDCl₃, 200 MHz)

<u>Compound</u>	<u>de %</u>	<u>CH₃</u>		<u>N-CH₃</u>	
		<u>major</u>	<u>minor</u>	<u>major</u>	<u>minor</u>
4a	66	0.86	0.71	2.22	2.40
4b	64	0.83	0.69	2.21	2.37
4c	66	0.85	0.72	2.21	2.37
4d	64	0.83	0.70	2.20	2.35
4e	66	0.84	0.72	2.19	2.34

synthesis of α -substituted phenethylamines 5 is based upon the double alkylation of the aminonitrile 1 followed by a stereoselective decyanation.

Thus, sequential dialkylation of 1 under the conditions previously described⁴ afforded 1,3-oxazolidines 3a-e (Scheme) that, without isolation were subjected to reductive decyanation in EtOH using NaBH₄ at room temperature. The separation of the diastereomers 3a-e was possible but not helpful since the decyanation reaction which is an elimination-addition process was proved to be non-stereospecific. Compounds 4a-e, isolated as viscous oils, consist of two diastereomers. The d.e.s. were measured by integration of the two methyl signals in the ¹H NMR spectra of the crude mixture (Table). In the course of this work no separation of the diastereomers 4a-e was achieved and the chiral appendage of 4a-e was removed by hydrogenolysis (Pd/C, 10 %, MeOH, rt, 12h) giving (S)-N-methyl- α -methyl-phenethylamines 5a-d⁶ in 56-62 % e.e. (Table). The e.e. values for compounds 5 were determined by polarimetry in the case of 5a and 5c and by the ¹H NMR spectra of the Mosher's amides⁷ in the case of 5b and 5d. The good agreement of d.e.s. for 4 and e.e. for 5 demonstrated that no racemisation occurred in the hydrogenolysis step.

In conclusion, the procedure described here should be valuable for the synthesis of various α -alkyl phenethylamines since alkylation of synthon 1 can be achieved with a series of alkyl and benzyl halides. Separation of the diastereomers 4 would lead after hydrogenolysis of the chiral auxiliary to optically pure R or S amphetamines 5.

EXPERIMENTAL SECTIONPreparation of 3a

The following procedure is typical of the experimental conditions used for the preparation of compounds 3. Compounds 3 are used without purification for the next step.

To a stirred solution of LDA/HMPA (1/1 ; $5 \cdot 10^{-3}$ mol, prepared from 3.12 mL of BuLi 1.6 M and 0.75 mL of diisopropylamine at $-10^{\circ}/-30^{\circ}$ C under a nitrogen atmosphere) in THF (20 mL) was added 2⁴ (0.909 g, $45 \cdot 10^{-4}$ mol, THF, -78° C) via syringe over 5 min. After 20 min. the resultant anion solution was added benzyl bromide (2 equiv.) and the mixture was stirred at -78° C for 1 h, quenched by NH_4Cl then extracted with ether dried and concentrated to dryness. Flash chromatography of the residual oil (SiO_2 , hexane-AcOEt, 80-20) yielded compound 3a as a mixture of diastereomers :

MS m/z (relative intensity) 292 (M^{+} , 1), 277 (66), 251 (18), 220 (13), 201 (100), 157 (25), 132 (26), 105 (54), 91 (74), 81 (56).

¹H NMR (CDCl_3 , 200 MHz) (ppm) :

major compound : 1.08 (s, 3H, CH_3), 2.89 (d, $J = 13.5$ Hz, 1H, CH_2 Ph), 3.13 (d, $J = 13.5$ Hz, 1H, CH_2 Ph), 4.86 (d, $J = 4$ Hz, 1H, N-CH-O), 5.02 (d, $J = 4$ Hz, 1H, N-CH-O)

minor compound : 1.34 (s, 3H, CH_3), 2.39 (d, $J = 13.5$ Hz, 1H, CH_2 Ph), 3.03 (d, $J = 13.5$ Hz, 1H, CH_2 Ph), 4.69 (d, $J = 3.5$ Hz, 1H, N-CH-O), 4.92 (d, $J = 3.5$ Hz, 1H, N-CH-O)

anal. calcd. for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{O}$: C, 77.52 ; H, 6.80 ; N, 9.52.

found : C, 77.64 ; H, 6.70 ; N, 9.55.

Preparation of 4

NaBH₄ (1.1 eq.) was added portionwise to a solution of 3 in EtOH (~ 0.4 M); stirring was continued at room temperature overnight.

The solvent was removed and the residue taken in CH₂Cl₂, washed with water then with brine, dried over Na₂SO₄ and evaporated to dryness. The oily residue was purified by flash-chromatography (CH₂Cl₂-MeOH : 95-5)

4a oil (30 % yield from 1)

MS (m/z, relative intensity) : 238 (M⁺ - 31, 24), 180 (34), 178 (100), 121 (24), 71 (34)

¹H NMR (see table)

anal. calcd. for C₁₈H₂₃NO : C, 77.32 ; H, 8.25 ; N, 5.01

found : C, 77.21 ; H, 8.17 ; N, 5.27

4b oil (29 % yield from 1)

MS (m/z, relative intensity) : 268 (M⁺ - 31, 4), 191 (8), 178 (100), 161 (9), 121 (35), 103 (14), 91 (16)

¹H NMR (see table)

anal. calcd. for C₁₉H₂₅NO₂ : C, 76.15 ; H, 8.35 ; N, 4.67

found : C, 76.08 ; H, 8.26 ; N, 4.52

4c oil (25 % yield from 1)

MS (m/z, relative intensity) : 268 (M⁺ - 31, 15), 178 (100), 149 (6), 121 (27), 103 (12), 91 (9)

¹H NMR (see table)

anal. calcd for C₁₉H₂₅NO₂ : C, 76.15 ; H, 8.35 ; N, 4.67

found : C, 76.09 ; H, 8.12 ; N, 4.57

4d oil (21 % yield from 1)

MS (m/z, relative intensity) : 252 (M^{+} - 31, 20), 178 (100), 121 (65), 105 (62), 103 (49), 91 (35), 77 (27)

^1H NMR (see table)

anal. calcd. for $\text{C}_{19}\text{H}_{25}\text{NO}$: C, 80.48 ; H, 8.82 ; N, 4.94

found : C, 80.72 ; H, 8.78 ; N, 4.87

4e oil (26 % yield from 1)

MS (m/z, relative intensity) : 272, 274 (M^{+} - 31, 14, 5), 178 (100), 125 (22), 121 (25), 103 (21), 91 (18), 77 (10), 58 (84)

^1H NMR (see table)

anal. calcd. for $\text{C}_{18}\text{H}_{28}\text{NOCl}$: C, 71.06 ; H, 7.23 ; N, 4.60

found : C, 70.86 ; H, 7.41 ; N, 4.55

Preparation of 5

Compound 4 dissolved in methanol (~ 0.2 M) was submitted to hydrogenolysis at room temperature and atmospheric pressure in presence of 10 % Pd/C overnight. The reaction mixture was then filtered through a celite bed and the filtrate evaporated in vacuo to yield an oily residue which was purified by flash chromatography on silica (CH_2Cl_2 -MeOH).

5a mp. 35-37° (solidified oil) Y : 90 %

^1H NMR (CDCl_3 , 80 MHz) δ (ppm) : 1.03 (d, $J = 7.5$ Hz, 3H, CH_3), 1.57 (s, 1H, NH), 2.35 (s, 3H, N- CH_3), 2.42-2.95 (m, 3H), 7.15 (s, 5H, Ar)

MS : m/z (relative intensity), 134 (M^{+} - 31, 10), 119 (5), 117 (5), 91 (45), 65 (16), 58 (100)

anal. calcd. for $\text{C}_{10}\text{H}_{15}\text{N}$: C, 80.53 ; H, 10.06 ;

found : C, 80.37 ; H, 9.73

5a-HCl mp 170° (EtOH) (lit^{2c} 172°)

$[\alpha]_D^{25}$ 12.0° (H₂O, c 1.2), lit^{2c} : $[\alpha]_D^{25}$ 19.3° (H₂O, c 1.2) ; e.e. 62 %

5b oil (Y : 55 %), $[\alpha]_D^{25}$ 1.2° (CHCl₃, c 1.6) ; e.e. 56 %

¹H NMR (CDCl₃, 80 MHz) δ (ppm) : 1.06 (d, J = 7Hz, 3H, CH₃), 2.16 (s, 1H, NH), 2.40 (s, 3H, N-CH₃), 2.47-2.95 (m, 3H), 3.77 (s, 3H, OCH₃), 6.82 (d, J = 8 Hz, 2H, ar.), 7.10 (d, J = 8 Hz, 2H, Ar)

MS : m/z (relative intensity), 179 (M⁺, 2), 178 (2), 164 (3), 121 (20), 72 (25), 58 (100)

5c oil (Y : 42 %), $[\alpha]_D^{25}$ 5.8° (CH₂Cl₂, c 1.98) lit^{2c} $[\alpha]_D^{23}$ 9.7° (CH₂Cl₂, c 2) ; e.e. 60 %

¹H NMR (CDCl₃, 80 MHz) δ (ppm) : 1.06 (d, J = 7 Hz, 3H, CH₃), 1.78 (s, 1H, NH), 2.38 (s, 3H, N-CH₃), 2.48-3.16 (m, 3H), 3.77 (s, 3H, OCH₃), 6.58-7.60 (m, 4H, Ar)

MS : m/z (relative intensity), 179 (M⁺, 5), 164 (6), 148 (13), 121 (17), 91 (12), 96 (7), 72 (11), 58 (100)

5d oil (Y : 45 %) $[\alpha]_D^{25}$ 2.7° (CHCl₃, c 0.6) ; e.e. 62 %

¹H NMR (CDCl₃, 80 MHz) δ (ppm), 1.05 (d, J = 7.5 Hz, 3H, CH₃), 1.37 (s, 1H, NH), 2.28 (s, 3H, Ar-CH₃), 2.37 (s, 3H, N-CH₃), 2.5-3.1 (m, 3H), 7.12 (m, 4H, Ar)

MS : m/z (relative intensity) 163 (M⁺, 2), 148 (3), 105 (6), 91 (4), 77 (4), 58 (100)

REFERENCES AND NOTES

- 1 - For part VIII see Bonin, M. ; Royer, J. ; Grierson, D.S. and Husson, H.-P., Tetrahedron Lett., 1986, 27, 1569.

- 2 - See for instance : a) Weinges, F. and Graab, G., Chem. Ztg. Chem. Appl., 1970, 94, 728. b) Nichols, D.E. ; Barfknecht, C.F. and Rusterholz, D.B., J. Med. Chem., 1973, 16, 480. c) Buckley III, T.F. and Rapoport, H., J. Am. Chem. Soc., 1981, 103, 6157. d) Chanut, F. ; Hoa, H.A. and Guette, J.P., Fourth European Symposium on Organic Chemistry, Poster Session, Aix-en-Provence (France), sept. 2-6, 1985.

- 3 - Takahashi, H. ; Chida, Y. ; Higashiyama, K. and Onishi, H., Chem. Pharm. Bull., 1985, 33, 4662.

- 4 - Marco, J.L. ; Royer, J. and Husson, H.-P., Tetrahedron Lett., 1985, 26, 3567.

- 5 - a) Grierson, D.S. ; Harris, M. and Husson, H.-P., J. Am. Chem. Soc., 1980, 102, 1064. b) Guerrier, L. ; Royer, J. ; Grierson, D.S. and Husson, H.-P., ibid., 1983, 105, 7754.

- 6 - In the case of 4e simultaneous aromatic dechlorination occurred and compound 5a was obtained in quantitative yield.

- 7 - Dale, J.A. and Mosher, H.S., J. Am. Chem. Soc., 1973, 95, 512.