

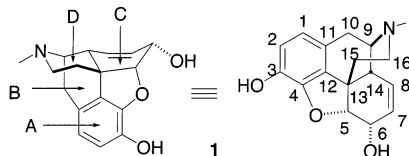
Synthesis of (–)-Morphine

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Morphine (**1**) is the principal alkaloid of opium, derived from *Papaver somniferum* L., or *P. album* Mill, *Papaveraceae*.¹ Morphine is also found in normal brain, blood, and liver tissue.² The morphine alkaloids comprise a family of structurally related natural products of unique clinical importance in medicine.³ The unusual architecture of morphine has offered a continuing challenge to the art and science of organic synthesis.^{4–6}



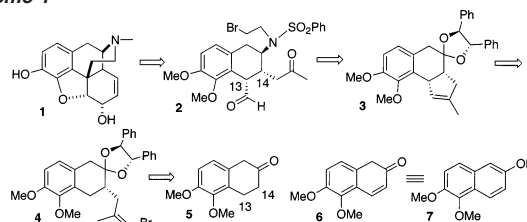
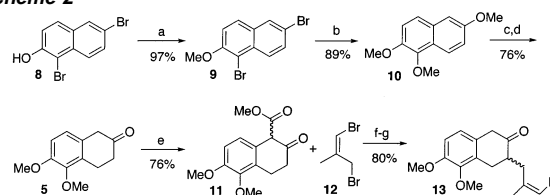
We envisioned that (–)-morphine **1** could ultimately be constructed from the easily prepared 5,6-dimethoxy- β -tetralone **5** (Scheme 1). A key step in this approach was the bis-intramolecular cyclization of the keto aldehyde **2**. The challenge was the introduction of the formyl substituent at C-13 (morphine numbering). Conjugate addition to an enone such as **6** would not be possible, as the enone **6** would tautomerize to the β -naphthol **7**. We hypothesized that initial alkylation of **5** at the C-14 position followed by ketalization with (*S,S*)-(–)-hydrobenzoin would give the bromoalkene **4**. Intramolecular alkylidene C–H insertion⁷ would then convert bromoalkene **4** to the cyclopentene **3**, and thus give access to **2**.

Our approach to the synthesis of (–)-morphine **1** began with the preparation of β -tetralone **13** (Scheme 2). Using modifications of the published procedures,⁸ we alkylated 1,6-dibromo-2-naphthol **8** with iodomethane to give the methoxynaphthalene **9**. Ullman coupling with sodium methoxide then gave the desired trimethoxynaphthalene **10**. Dissolving metal reduction followed by hydrolysis led to the desired β -tetralone **5**. The β -tetralone **5** would tend to alkylate at the benzylic position. The procedure of Aristoff,⁹ methoxycarbonylation, dianion alkylation using *cis*-1,3-dibromo-2-methyl-1-propene,¹⁰ and decarboxylation, was therefore employed to obtain the alkylated β -tetralone **13**.

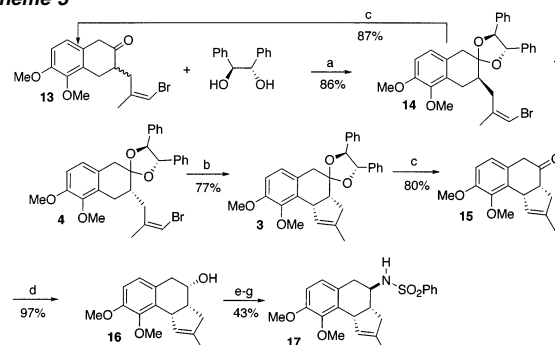
Protection of the β -tetralone **13** (Scheme 3) with (*S,S*)-(–)-hydrobenzoin gave the diastereomeric ketals **14** and **4**, which, as anticipated, were separable by silica gel chromatography. The undesired diastereomer **14** was readily recycled to the racemic β -tetralone **13**. Cyclization of ketal **4** via alkylidene carbene C–H insertion⁷ followed by hydrolysis led to the enantiomerically pure ketone **15**. The beauty of this approach is that while β -tetralone **13** can readily racemize, β -tetralone **15** cannot.

The sterically congested ketone **15** was selectively reduced to the *cis* alcohol **16**. Direct displacement of the alcohol by a functionalized amine could not be achieved. Fortunately, the alcohol

Scheme 1

Scheme 2^a

^a Conditions: (a) CH₃I, K₂CO₃, DMF; (b) NaOCH₃, collidine, CuI, MeOH, reflux; (c) Na, EtOH, reflux; (d) HCl, H₂O, reflux; (e) (CH₃O)₂CO, NaOMe, MeOH, reflux; (f) LDA (2 equiv), THF, 0 °C; (g) LiCl, DMSO, H₂O, reflux.

Scheme 3^a

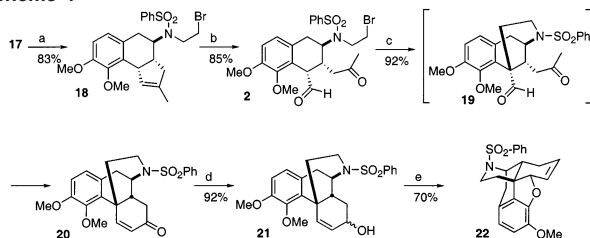
^a Conditions: (a) *p*-TSA, HC(OEt)₃, CH₂Cl₂; (b) KHMDS, Et₂O; (c) AcOH, H₂O, reflux; (d) L-selectride, THF, 0 °C; (e) (PhO)₂P(O)N₃, DEAD, Ph₃P, THF; (f) LAH/EtOH – (1/1), Et₂O; (g) PhSO₂Cl, Et₃N, CH₂Cl₂.

16 was smoothly converted to the azide via Mitsunobu coupling. Reduction and protection then gave sulfonamide **17**.

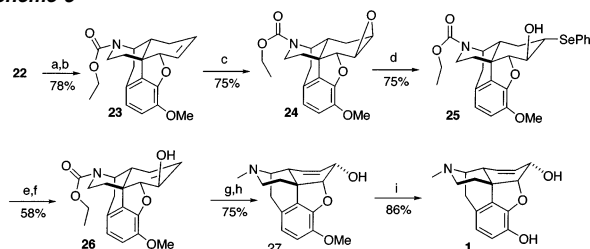
The key to the assembly of morphine was the anticipated selective bis-cyclization of keto aldehyde **2** (Scheme 4). Alkylation of the sulfonamide **17** with 1,2-dibromoethane under phase-transfer conditions provided **18**, which upon ozonolysis gave the desired keto aldehyde **2**. The benzylic proton α to the aldehyde in **2** is the most acidic, so we expected to obtain the aldehyde enolate selectively. Although the keto aldehyde **19** could be isolated after brief exposure to base, it was more practical to continue heating, to cleanly obtain the tetracycle **20**. The final conversion to complete the core structure of morphine **1** was the construction of the ether ring. Reduction of the enone **20** gave a single alcohol **21** (Scheme

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[§] For X-ray analysis.

Scheme 4^a

^a Conditions: (a) BrCH₂CH₂Br, 1 N NaOH, TBAB, toluene, reflux; (b) O₃, CH₂Cl₂, -78 °C, Ph₃P; (c) K₂CO₃, TBAB, toluene, reflux; (d) NaBH₄, EtOH; (e) BBr₃, CH₂Cl₂, -40 °C.

Scheme 5^a

^a Conditions: (a) Red-Al, toluene, reflux; (b) ClCOOEt, Et₃N, CH₂Cl₂; (c) [(C₈H₁₇)₃NCH₃]⁺₃[PO₄[W(O)(O₂)₂]₄]³⁻, H₂O₂, DCE, reflux; (d) Ph-SeSePh, NaBH₄, EtOH, reflux; (e) NaIO₄, THF, H₂O; (f) Na₂CO₃, toluene, H₂O; (g) MnO₂, CH₂Cl₂; (h) LiAlH₄, THF, reflux; (i) BBr₃.

4), which upon brief exposure to BBr₃ gave clean cyclization to **22**, having the pentacyclic morphine skeleton.

The next challenge (Scheme 5) was the removal of the robust phenylsulfonyl protecting group. Although dissolving metal conditions failed, we found that Red-Al was very effective¹¹ for this difficult deprotection. Reprotection immediately followed to give the carbamate **23**.

To effect the final oxidation to the allylic alcohol of morphine, we first epoxidized the alkene **23** with H₂O₂.¹² Regioselective ring opening of the epoxide **24** then gave the selenide **25**. The expected selectivity exhibited in both the epoxidation and the epoxide opening was controlled by the strong steric influence of the arene ring, which effectively blocks both the lower face of the C ring and the backside attack at the C-6 position. Oxidation of the selenide **25** followed by elimination yielded the allylic alcohol **26** with the configuration at C-6 opposite to that of morphine. Manganese dioxide oxidation followed by LiAlH₄ reduction proceeded with the reported¹³ high diastereocontrol to deliver codeine **27**. Finally, O-demethylation¹⁴ gave morphine **1**, identical (TLC, ¹H NMR, ¹³C NMR, [α]_D) with natural material.

A β-tetralone-based approach to the synthesis of (-)-morphine **1** has been achieved, in 23 steps from **5**, with an overall yield of 0.77%. This synthesis opens the way to the preparation of a variety of C-10, C-15, and C-16 substituted morphine analogues that have previously not been available. The strategy outlined here for the enantioselective construction of three contiguous stereogenic centers and the novel ring cyclizations that followed will have many applications in target-directed organic synthesis.

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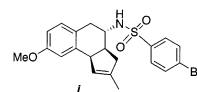
spectroscopy, and to Kenner C. Rice for many helpful discussions. This work is dedicated to the memory of Henry Rapoport and Arthur G. Schultz, masters of the science and art of alkaloid synthesis.

Note Added after Print Publication: Due to a production error, the graphics were incomplete in the version published on the Web 09/28/2002 (ASAP) and in the October 23, 2002 issue (Vol. 124, No. 42, pp 12416–12417); the correct electronic version of the paper was published on 11/27/02 and an Addition and Correction appears in the December 25, 2002 issue (Vol. 124, No. 51).

Supporting Information Available: Details for the preparation of compounds **1–27** (PDF), and X-ray data for compound *i* (CIF).¹⁵ This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (15) β-Tetralone **13** was initially ketalized with (R,R)-(+)-hydrobenzoin. The first ketal diastereomer to elute via chromatography was converted to the p-bromobenzenesulfonamide *i*. This was determined by X-ray analysis to have the configuration at C-9, C-13, and C-14 shown.



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