

The Efficient Synthesis of the Optically Active β -Hydroxyl- γ -butyrolactone Derivatives

Jin Xin Wang (王進欣), Ying Li* (李瀛) and Chao Xin Zhang (張朝欣)

National Laboratory of Applied Organic Chemistry, Lanzhou University, Lanzhou, 730000, P. R. China

The optically active β -hydroxyl- γ -butyrolactones were synthesized from nonchiral starting material by employing reductive cleavage reaction, Sharpless asymmetric epoxidation and dihydroxylation, and Lewis acid-catalysed cyclization as key steps. This strategy can be used to prepare many chiral β -hydroxyl- γ -butyrolactone analogues.

Keywords: Asymmetric synthesis; β -Hydroxy- γ -butyrolactone; Reductive cleavage reaction; Sharpless epoxidation and dihydroxylation; Cyclization.

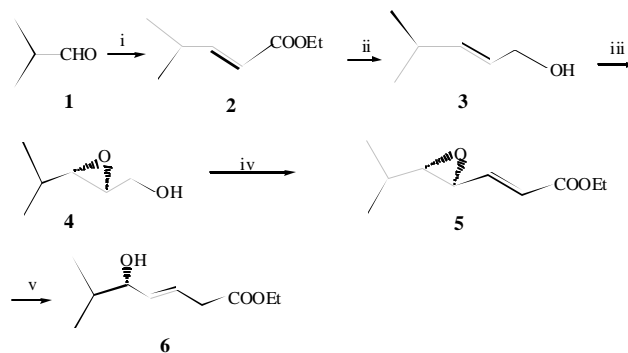
INTRODUCTION

Chiral γ -butyrolactones have attracted substantial interest in recent years due to their presence in many strongly active natural products having antitumor, fungicidal, anti-inflammatory activity,^{1,2,3} and their use as important precursors in natural product synthesis.⁴ Presently the asymmetric synthesis of β -hydroxy- γ -butyrolactones have been a target for organic synthesis in various laboratories. In the course of the total synthesis of the natural product Tuxpanolide⁵ and its analogues,⁶ we prepared the various optically active β -hydroxyl- γ -butyrolactones derivatives as needed key intermediates, finding a practical strategy for building chiral β -hydroxy- γ -butyrolactones. In this paper, we present our results on the efficient stereocontrolled synthesis of β -hydroxyl- γ -butyrolactone derivatives from cheap and nonchiral starting material.

RESULTS AND DISCUSSION

The starting material isobutyraldehyde **1** was subjected to the Wittig reaction followed by the reduction with $\text{LiAlH}_4\text{-AlCl}_3$ in dry ether to give the allylic alcohol **3**. Sharpless asymmetric epoxidation⁷ of the allylic alcohol **3** with (+)-DET, $\text{Ti}(i\text{-OPr})_4$, TBHP led to the epoxy alcohol (+)-**4** in 71% yield⁸ and 90% ee as determined by $^1\text{H NMR}$ analysis of the corresponding Mosher's ester.⁹ Coupling reaction of aldehyde obtained by Swern oxidation of **4** and triphenylphosphorane afforded (+)-**5** (41% for two steps). After the synthesis of four steps, by simply applying the reductive cleavage reaction of α,β -unsaturated ester **5** by magnesium in methanol,¹⁰ we luckily accessed the sole product (+)-**6** in 72% yield (Scheme I).

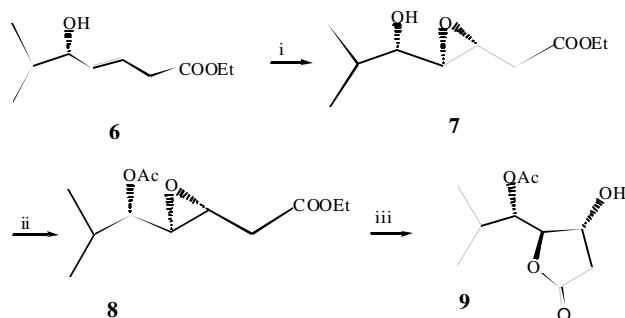
Scheme I



(i) $\text{Ph}_3\text{P}=\text{CHCOOEt}$, CH_2Cl_2 , 0°C -r.t.; (ii) $\text{LiAlH}_4\text{:AlCl}_3 = 3\text{:}1$, Et_2O , -78°C ; (iii) $\text{Ti}(\text{O-}i\text{-Pr})_4$, (+)-DET, TBHP, CH_2Cl_2 , -20°C ; (iv) (a) $(\text{COCl})_2$, DMSO, Et_3N , CH_2Cl_2 , -78°C ; (b) $\text{Ph}_3\text{P}=\text{CHCOOEt}$, CH_2Cl_2 , r.t.; (v) Mg , MeOH, -23°C .

Subsequent treatment of **6** with $t\text{-BuOOH-VO}(\text{acac})_2$ catalyzed system in dry benzene gave a 7.5:1 mixture (determined by GLC, 88% de) of the desired α -epoxy alcohol (+)-**7** and its β -isomers in 61% yield. After the protection of a secondary hydroxy of **7**, the (+)-**8** was obtained in 73% yield. Finally with the treatment of **8** with camphorsulfonic acid (CSA) catalysed cyclization,¹¹ we proceeded to construct a chiral butyrolactone (+)-**9** in 63% yield, 98% de determined by GLC, and 95% ee in accordance with the $^1\text{H NMR}$ analysis of the corresponding Mosher's ester⁹ (Scheme II). This route successfully employed the Sharpless catalytic asymmetric epoxidation reaction, which allowed the two chiral centers of the intermediate **3** to remain *S* in a highly predictable way. Moreover, the configurations of C_3 , C_4 in the compound **7** were determined by the fact that the vanadium-catalyzed epoxidation exhibited the cis stereoselectivity.¹² It provided

Scheme II

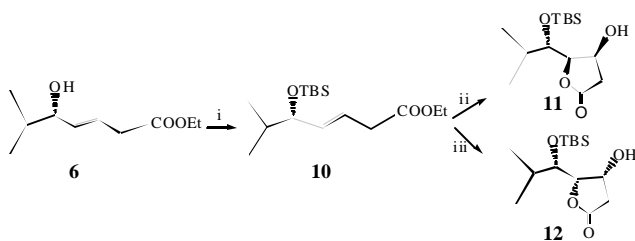


(i) *t*-BuOOH, VO(acac)₂, benzene, 5 °C-r.t. (ii) Ac₂O, DMAP, Py, r.t.; (iii) CSA, CH₂Cl₂, 0 °C-r.t.

the efficient stereocontrolled synthetic approach for building the trans β-hydroxy-γ-butyrolactone blocks.

The Sharpless asymmetric dihydroxylation (ADs) of olefins is an indispensable tool for contemporary organic synthesis. In Scheme III, the key intermediate **6** was allowed to react with *tert*-butyldimethylchlorosilane (1.2 eq) and imidazole (3.0 eq) in anhydrous DMF. After 6 hours, **10** was obtained in 82% yield. Then ADs of **10** (in accordance with the literature precedence¹³) provided lactonized dihydroxylation products **11** (ee 73%) and **12** (ee 78%), respectively, in 45% and 51% yield. Thus, the *cis* β-hydroxy-γ-butyrolactones **11** and **12** were successfully obtained by this economical and efficient method. Remarkably, if **6** was protected with an acetyl, the AD reaction did not happen.

Scheme III



(i) TBDMCl, Et₃N, DMAP, DMF, r.t.; (ii) AD-mix-β, CH₃SO₂NH₂, *t*-BuOH-H₂O, r.t.; (iii) AD-mix-α, CH₃SO₂NH₂, *t*-BuOH-H₂O, r.t.

Here we report the asymmetric synthesis of the optically active β-hydroxy-γ-butyrolactone derivatives **9**, **11**, **12**. With the above Lewis acid catalyzed cyclization and ADs lactonization, the various chiral β-hydroxy-γ-butyrolactones can be constructed generally and practically via the key intermediate **6** and its derivatives.

EXPERIMENTAL SECTION

General Methods

IR spectra were recorded on an FT-170SX spectrometer. ¹H NMR and ¹³C NMR spectra were recorded on Bruker AM-200 or AM-400 MHz instruments using tetramethylsilane (TMS) as the internal standard. Mass spectra were recorded on VG ZAB-HS or VG-7070 (70 eV) spectrometers. GLC analyses were carried out on a Shimadzu GC-9AM instrument. Optical rotations were measured with a Perkin Elmer 341 instrument.

(*E*)-4-Methyl-2-pentenoate (**2**)

To a stirred solution of (ethoxycarbonylmethylene) triphenylphosphorane (20 g, 57.4 mmol) in dry CH₂Cl₂ (50 mL) was added isobutyraldehyde (4.14 g, 57.5 mmol) under an Ar atmosphere. The reaction mixture was stirred for 2 h at room temperature and the solvent was evaporated. The crude residue was purified by column chromatography (petroleum) to furnish **2** as a colorless liquid (6.5 g, 80%). IR (film): ν = 2982, 1708, 1651, 1514, 1436, 1314, 1290, 1207, 1150, 981 cm⁻¹. ¹H NMR (200 M, CDCl₃): δ 1.05 (d, *J* = 7.0 Hz, 6H), 1.23 (t, *J* = 7.2 Hz, 3H), 2.38-2.42 (m, 1H), 4.12 (q, *J* = 7.2 Hz, 2H), 5.72 (d, *J* = 15.8 Hz, 1H), 6.90 (dd, *J* = 15.8, 6.6 Hz, 1H). The data matched those reported in the literature.¹⁴

(*E*)-4-Methyl-2-penten-1-ol (**3**)

To a stirred and precooled (-78 °C) solution of LiAlH₄ (294 mg, 7.75 mmol) and AlCl₃ (344 mg, 2.58 mmol) in dry ether (25 mL) was added dropwise compound **2** (366 mg, 2.58 mmol). The reaction was stirred for 1.5 h at -78 °C and then H₂O was added slowly to quench the unreacted LiAlH₄. The mixture was filtered and the ether solution was washed by water and brine. The etheral layer was dried over MgSO₄. Removal of solvent by rotary evaporation yielded a colorless liquid **3** (237 mg, 92%). IR (film): ν = 3350, 2934, 2889, 1117, 1465, 1382, 1205, 980 cm⁻¹. ¹H NMR (200 M, CDCl₃): δ 0.94 (d, *J* = 6.6 Hz, 6H), 2.22-2.27 (m, 1H), 2.43 (brs, OH), 4.01 (d, *J* = 5.6 Hz, 2H), 5.48-5.55 (m, 1H), 5.6 (dd, *J* = 15.7 Hz, 6.0 Hz, 1H). EIMS: *m/e* 101, 81, 55, 43. The data were consistent with those reported in the literature.¹⁵

(+)-(4*S*,5*S*)-6-Methyl-4,5-epoxy-2-heptenoate (**5**)

To a stirred solution of oxalyl chloride (0.23 mL, 2.6 mmol) in CH₂Cl₂ (10 mL) at -78 °C was added dropwise DMSO (0.37 mL, 5.2 mmol). Upon complete addition, **4** (150 mg, 1.3 mmol) dissolved in CH₂Cl₂ (0.5 mL) was added dropwise. The initially clear solution became white and cloudy after stirring for 1.5 h. Triethylamine (657 mg, 6.5

mmol) was then added dropwise at $-78\text{ }^{\circ}\text{C}$. Then the reaction mixture was warmed slowly to $-10\text{ }^{\circ}\text{C}$ for 2 h and quenched by addition of water (0.3 mL). The organic layer was separated and washed with water and brine; the combined aqueous washes were extracted with CH_2Cl_2 . The organic phases were combined and dried over MgSO_4 . After the removal of solvent, buff oil (100 mg) was obtained. To a stirred solution of (ethoxycarbonylmethylene) triphenylphosphorane (306 mg, 0.89 mmol) in dry CH_2Cl_2 (10 mL) was added to the buff oil (100 mg) under an Ar atmosphere. The reaction mixture was stirred for 2 h at room temperature and the solvent was evaporated. The crude residue was purified by column chromatography (petroleum:EtOAc, 32:1) to furnish **5** as a buff oil (110 mg, two steps 41%). $[\alpha]_{\text{D}}^{27} 10.6^{\circ}$ (*c* 1.4 CH_2Cl_2). IR (film): $\nu = 2962, 1725, 1661, 1593, 1439, 1307, 1275, 1192, 978, 857\text{ cm}^{-1}$. $^1\text{H NMR}$ (200 M, CDCl_3): δ 0.96 (d, *J* = 7.1 Hz, 3H), 1.01 (d, *J* = 6.8 Hz, 3H), 1.27 (t, *J* = 7.0 Hz, 3H), 1.56-1.64 (m, 1H), 2.67 (dd, *J* = 1.8, 6.6 Hz, 1H), 3.25 (dd, *J* = 6.8, 1.9 Hz, 1H), 4.22 (q, *J* = 7.0 Hz, 2H), 6.11 (d, *J* = 15.6 Hz, 1H), 6.69 (dd, *J* = 15.6, 6.7 Hz, 1H). $^{13}\text{C NMR}$ (50 M, CDCl_3): δ 14.07, 18.04, 18.74, 30.41, 55.09, 60.38, 66.50, 123.26, 144.79, 165.55. EIMS: *m/e* 185, 139, 111, 98, 83, 56, 45, 43, 41.

(+)-(5S)-6-Methyl-5-hydroxy-3-heptenoate (6)

The substrate **5** (160 mg, 0.87 mmol) in dry methanol (5 mL) was cooled at $-23\text{ }^{\circ}\text{C}$ before magnesium powder (63 mg, 2.61 mmol) was added. The reaction mixture was stirred for 2 h under Ar atmosphere. To the gray solution was added an equal volume of diethyl ether, the whole mixture was filtered through a silica gel pad and concentrated in vacuo, and crude product was purified by flash chromatography (SiO_2) (petroleum:EtOAc, 8:1) to obtain a colorless oil **6** (11 mg, 72%). $[\alpha]_{\text{D}}^{26} +12.1^{\circ}$ (*c* 1.8 CH_2Cl_2). IR (film): $\nu = 3410, 2922, 1733, 1626, 1405, 1381, 1158, 1072, 1023\text{ cm}^{-1}$. $^1\text{H NMR}$ (200 M, CDCl_3): δ 0.87 (d, *J* = 2.4 Hz, 3H), 0.90 (d, *J* = 2.4 Hz, 3H), 1.22 (t, *J* = 7.2 Hz, 3H), 1.63-1.70 (m, 1H), 3.05 (brd, 2H), 3.88 (t, *J* = 6.4 Hz, 1H), 4.13 (q, *J* = 7.2 Hz, 2H), 5.55 (dd, *J* = 7.2, 15.4 Hz, 1H), 5.73 (dt, *J* = 7.5, 15.4 Hz, 1H). $^{13}\text{C NMR}$ (50 M, CDCl_3): δ 14.11, 17.88, 18.10, 33.64, 37.69, 60.63, 72.33, 123.80, 135.30, 171.65. EIMS: *m/e* 143, 130, 97, 73, 43. HRMS ($\text{M}+\text{NH}_4$) calcd for $\text{C}_{10}\text{H}_{22}\text{O}_3\text{N}$ 204.1594, found 204.1597.

(+)-(3R,4S,5S)-6-Methyl-5-hydroxy-3,4-epoxy-heptanoate (7)

To a stirred solution of **6** (136 mg, 0.73 mmol) in anhydrous benzene (5 mL) at $5\text{ }^{\circ}\text{C}$ under Ar atmosphere, was added $\text{VO}(\text{acac})_2$ 10 mg (0.04 mmol). After the reaction mixture

was stirred for 10 minutes at $5\text{ }^{\circ}\text{C}$, *tert*-butylhydroperoxide 0.5 mL (1.46 mmol) was added dropwise and the resulting mixture was stirred for 12 h at the room temperature. The quenching was made by adding a saturated solution of NaHCO_3 , followed by extraction with benzene. Then the organic layer was washed with saturated solution of $\text{Na}_2\text{S}_2\text{O}_3$, water and then brine, and dried over MgSO_4 . After evaporation of the solvent, the residue was subjected to silica gel column chromatography (petroleum:EtOAc, 4:1) to give a 7.5:1 mixture of α -epoxide **7** (79 mg, 54%, 88% de) and β -epoxides (11 mg, 7%, 12% de) as a colorless oil. Compound **7**: $[\alpha]_{\text{D}}^{26} +8.0^{\circ}$ (*c* 2.0 CH_2Cl_2). IR (film): $\nu = 414, 2971, 1733, 1467, 1373, 1235, 1026\text{ cm}^{-1}$. $^1\text{H NMR}$ (200 M, CDCl_3): δ 0.96 (d, *J* = 2.4 Hz, 3H), 1.01 (d, *J* = 2.4 Hz, 3H), 1.31 (t, *J* = 7.6 Hz, 3H), 1.60-1.71 (m, 1H), 2.52 (dd, *J* = 5.6, 10.3 Hz, 1H), 2.63 (dd, *J* = 5.8, 10.4 Hz, 1H), 2.89 (dd, *J* = 2.2, 5.0 Hz, 1H), 3.30 (ddd, *J* = 2.2, 6.0, 11.0 Hz, 1H), 3.74 (dd, *J* = 5.0, 6.6 Hz, 1H), 4.20 (q, *J* = 7.6 Hz, 2H). $^{13}\text{C NMR}$ (50 M, CDCl_3): δ 14.14, 18.28, 19.55, 29.91, 32.37, 37.04, 52.18, 60.47, 71.23, 169.53. EIMS: *m/e* 159, 131, 117, 85, 71, 43. HRMS ($\text{M}+\text{NH}_4$) calcd for $\text{C}_{10}\text{H}_{22}\text{O}_4\text{N}$ 220.1543, found 220.1539.

(+)-(3R,4S,5S)-6-Methyl-5-acetoxy-3,4-epoxy-heptanoate (8)

To a solution of **7** (83 mg, 0.41 mmol) in pyridine (1 mL) was added acid anhydride (63 mg, 0.62 mmol) and DMAP (3 mg, 0.02 mmol); the reaction mixture was stirred for 12 h at room temperature. The mixture was extracted with EtOAc and then the organic layer was washed with aqueous 10% NaOH, 5% HCl, H_2O and brine, respectively, then dried over MgSO_4 . After removal of the solvent, the residue was subjected to silica gel column chromatography (petroleum:EtOAc, 8:1) to finish **8** (74 mg, 73%) as a buff oil. $[\alpha]_{\text{D}}^{26} +11.2^{\circ}$ (*c* 1.1 CH_2Cl_2). IR (film): $\nu = 2971, 2925, 1739, 1467, 1373, 1235, 1183, 1026\text{ cm}^{-1}$. $^1\text{H NMR}$ (400 M, CDCl_3): δ 0.96 (d, *J* = 4.7 Hz, 3H), 0.98 (d, *J* = 4.7 Hz, 3H), 1.25 (t, *J* = 7.4 Hz, 3H), 1.96-2.02 (m, 1H), 2.09 (s, 3H), 2.52 (dd, *J* = 5.3, 10.3 Hz, 1H), 2.63 (dd, *J* = 5.8, 11.0 Hz, 1H), 2.94 (dd, *J* = 1.8, 6.5 Hz, 1H), 3.21 (ddd, *J* = 1.8, 6.0, 11.0 Hz, 1H), 4.17 (q, *J* = 7.4 Hz, 2H), 4.53 (t, *J* = 6.6 Hz, 1H). $^{13}\text{C NMR}$ (100 M, CDCl_3): δ 14.06, 18.25, 18.60, 20.80, 28.96, 37.08, 52.53, 57.83, 60.88, 75.95, 169.73, 170.22. EIMS: *m/e* 201, 159, 131, 117, 85, 71, 43.

(+)-(3R,4R,5S)-4-(1-acetoxy-isopropyl)-3-Hydroxy-butyrolactone (9)

To a stirred solution of **8** (109 mg, 0.45 mmol) in anhydrous CH_2Cl_2 (3 mL) at $0\text{ }^{\circ}\text{C}$ under Ar atmosphere, was added camphorsulfonic acid (11 mg, 0.045 mmol). The reaction mixture

ture was stirred for 24 h at room temperature. The quenching was made by adding a saturated solution of NaHCO₃, followed by extraction with CH₂Cl₂, dried over MgSO₄ and concentrated to give the crude product purified by silica gel column chromatography (petroleum:EtOAc, 4:1), yielded a colorless gum **9** (56 mg, 63%, 99% de). [α]_D²⁶ +5.6° (*c* 0.7 CH₂Cl₂). IR (film): ν = 3414, 2964, 1783, 1743, 1378, 1235, 1178, 1073 cm⁻¹. ¹H NMR (400 M, CDCl₃): δ 97 (d, *J* = 6.9 Hz, 3H), 1.03 (d, *J* = 6.8 Hz, 3H), 1.70-1.95 (m, 1H), 2.11 (s, 3H), 2.49 (dd, *J* = 1.6, 18.6 Hz, 1H), 3.07 (dd, *J* = 7.2, 18.5 Hz, 1H), 3.51 (dd, *J* = 3.7, 8.0 Hz, 1H), 4.58 (dd, *J* = 1.7, 3.7 Hz, 1H), 5.45 (dd, *J* = 1.6, 7.2 Hz). ¹³C NMR (100 M, CDCl₃): δ 18.16, 18.87, 29.92, 35.92, 70.34, 85.84, 170.19, 175.17. EIMS: *m/e* 173, 155, 143, 84, 43. HRMS (M+NH₄) calcd for C₁₀H₂₂O₅N 234.1336, found 234.1336.

(+)-(5S)-6-Methyl-5-tert-butylidimethyl-siloxy-3-heptenoate (10)

To a solution of **6** (350 mg, 1.88 mmol) in anhydrous DMF (3 mL) at r.t. under Ar atmosphere, were added Et₃N (1.44 mL), TBDMCl (340 mg, 2.26 mmol) and DMAP (12 mg, 0.094 mmol). The mixture was stirred 6 hours at r.t., quenched with a saturated solution of NH₄Cl (1.5 mL) and diluted with Et₂O. The organic layer was washed with H₂O and brine, dried over MgSO₄ and concentrated to give the crude product purified by silica gel column chromatography (petroleum:EtOAc, 16:1), yielded a colorless oil **10** (463 mg, 82%). [α]_D²⁰ +9.6° (*c* 1.7 CH₂Cl₂). IR (film): ν = 2922, 1733, 1626, 1405, 1381, 1158, 1072, 1023 cm⁻¹. ¹H NMR (200 M, CDCl₃): δ 0.04 (s, 3H), 0.08 (s, 3H), 0.83 (brs, 3H), 0.86 (brs, 3H), 0.90 (s, 12H), 1.22 (t, *J* = 7.2 Hz, 3H), 1.63-1.70 (m, 1H), 3.04-3.09 (m, 2H), 3.80 (t, *J* = 6.4 Hz, 1H), 4.16 (q, *J* = 7.2 Hz, 2H), 5.54-5.62 (m, 2H). EIMS: *m/e* 243, 213, 185, 143, 130, 117, 97, 73, 43.

(+)-(3S,4R,5S)-4-(1-tert-butylidimethylsiloxy-isopropyl)-3-hydroxy-butyrolactone (11)

To compound **10** (208 mg, 0.68 mmol) was added a mixture of t-BuOH (3.5 mL), H₂O (3.5 mL), AD-mix β (966 mg, 0.69 mmol), and methanesulfonyl amide (66 mg, 0.69 mmol). The solution was stirred for 72 hours at 0 °C. After the addition of saturated Na₂SO₃ (1.0 g, 0.69 mmol), the mixture was stirred for 40 minutes and diluted with EtOAc. The organic layer was washed with H₂O and brine, dried over MgSO₄ and concentrated to give the crude product purified by silica gel column chromatography (petroleum:EtOAc, 4:1), yielded a colorless flake crystal **11** (82 mg, 45%). [α]_D²⁰ +7.0° (*c* 1.9 CH₂Cl₂). IR (film): ν = 3467, 2956, 2930, 2856, 1770, 1754, 1469, 1389, 1252, 1128, 1065, 841 cm⁻¹. ¹H

NMR (400 M, CDCl₃): δ 0.14 (s, 3H), 0.20 (s, 3H), 0.91 (s, 12H), 1.04 (d, *J* = 4.2 Hz, 3H), 1.06 (d, *J* = 4.2 Hz, 3H), 2.03-2.08 (m, 1H), 2.59 (brd, 1H), 2.69 (dd, *J* = 4.8, 17.4 Hz, 1H), 4.07 (dd, *J* = 1.9, 7.0 Hz, 1H), 4.33-4.35 (m, 1H), 4.71 (dd, *J* = 1.8, 7.7 Hz, 1H), 4.97 (d, *J* = 3.2 Hz, OH). ¹³C NMR (100 M, CDCl₃): δ 18.10, 19.02, 19.19, 25.82, 31.48, 40.48, 69.99, 81.31, 175.33. EIMS: *m/e* 287, 245, 231, 201, 187, 171, 159, 147, 129, 117, 113, 101, 25, 43. HRMS (M+NH₄) calcd for C₁₄H₃₂O₄NSi 306.2095, found 306.2099.

(-)-(3R,4S,5S)-4-(1-tert-butylidimethylsiloxy-isopropyl)-3-hydroxy-butyrolactone (12)

Compound **12** was prepared in the same method as **11**, obtained a colorless flake crystal in yield 51%. [α]_D²⁰ -14.0° (*c* 1.0 CH₂Cl₂). ¹H NMR (200 M, CDCl₃): δ 0.13 (s, 3H), 0.19 (s, 3H), 0.90 (s, 12H), 1.04 (d, *J* = 4.2 Hz, 3H), 1.06 (d, *J* = 4.2 Hz, 3H), 1.99-2.10 (m, 1H), 2.61-2.69 (m, 2H), 4.07 (dd, *J* = 3.0, 7.0 Hz, 1H), 4.31-4.35 (m, 1H), 4.69-4.71 (m, 1H), 4.96 (d, *J* = 3.2 Hz, OH). ¹³C NMR (50 M, CDCl₃): δ 18.10, 19.02, 19.22, 25.82, 31.49, 40.48, 70.01, 81.29, 175.33. IR, EIMS and HRMS were same as compound **11**.

ACKNOWLEDGMENT

We thank the Natural Science Foundation of China (20272020) and Special Research Grant for Doctor Sites in Chinese University.

Received January 2, 2003.

REFERENCES

- (a) Schelewer, G.; Stampf, J. L.; Benezra, C. *J. Med. Chem.* **1980**, *23*, 1031. (b) Niwa, M.; Iguchi, M.; Yamamura, S. *Tetrahedron Lett.* **1975**, *49*, 4395. (c) Papathanasopoulos, N.; Lazarus, H.; Foley, G. E.; Modest, E. J. *J. Med. Chem.* **1974**, *17*, 672. (d) Lee, K. H.; Ibuka, R. Y.; Geissman, T. A. *Phytochemistry* **1977**, *16*, 1177. (e) Kupchan, S. M.; Eakin, M. A.; Thomas, A. M. *J. Med. Chem.* **1971**, *14*, 1147. (f) Nair, V.; Sinhababu, A. K. *J. Org. Chem.* **1980**, *45*, 1893.
- Slob, A.; Jekel, B.; Jong, B. *Phytochemistry* **1975**, *14*, 1997.
- (a) Niwa, M.; Iguchi, M.; Yamamura, S. *Tetrahedron Lett.* **1975**, *49*, 1539. (b) Takeda, K.; Sakurawi, K.; Ishii, H. *Tetrahedron* **1972**, *28*, 3757. (c) Martines, J. C. V.; Yoshida, M.; Gottlieb, O. R. *Tetrahedron Lett.* **1979**, *12*, 1021. (d) Barbier, P.; Benezra, C. *J. Org. Chem.* **1983**, *48*, 2705. (e) Rollinson, S. W.; Amos, R. A.; Katzenellenbogen, J. A. *J.*

- Am. Chem. Soc.* **1981**, 103, 4114.
- (a) Koch, S. S. C.; Chamberlin, A. R. *In Enantiomerically Pure γ -Butyrolactones in Natural Products Synthesis*; Atta-ur-Rahman, Ed.; Elsevier Science: Amsterdam, 1995; pp 686-725. (b) Xu, M. H.; Wang, W.; Xia, L. J.; Lin, G. Q. *J. Org. Chem.* **2001**, 66, 3953.
 - Maldonado, E.; Bello, M.; Villasenor, J. L.; Ortega, A. *Phytochemistry* **1998**, 49, 1115.
 - Wang, J. X.; Zheng, S. Z.; Sun, L. P.; Shen, X. W.; Li, Y. *J. Chin. Chem. Soc.* **2002**, 49, 437.
 - Katsuki, T.; Sharpless, K. B. *J. Am. Chem. Soc.* **1980**, 102, 5974.
 - Caldwell, C. G.; Bondy, S. S. *Synthesis* **1990**, 1, 34.
 - (a) Hassner, A.; Alexanian, V. *Tetrahedron Lett.* **1978**, 46, 4475. (b) Hagiwara, H.; Uda, H. *J. Org. Chem.* **1988**, 53, 2308.
 - Pak, C. S.; Lee, E.; Lee, G. H. *J. Org. Chem.* **1993**, 58, 1523.
 - (a) Nacro, K.; Baltas, M.; Escudier, J. M.; Gorrichon, L. *Tetrahedron* **1997**, 53, 659. (b) Nicolaou, K. C.; Prasad, C. V. C.; Somers, P. K.; Hwang, C. K. *J. Am. Chem. Soc.* **1989**, 111, 5330.
 - (a) Henbest, H. B.; Wilson, R. A. *J. Chem. Soc.* **1959**, 81, 1958. (b) Itoh, T.; Jitsukawa, K.; Kaneda, K.; Teranishi, S. *J. Am. Chem. Soc.* **1979**, 101, 162.
 - Wang, Z. M.; Zhang, X. L.; Sharpless, K. B. *Tetrahedron Lett.* **1992**, 33, 6407.
 - Marmor, R. S. *J. Org. Chem.* **1972**, 37, 2901.
 - Gorthey, L. A.; Vairamani, M.; Djerassi, C. *J. Org. Chem.* **1984**, 49, 1511.